Journal of Zhejiang University-SCIENCE A (Applied Physics & Engineering)

<u>Cite this as:</u> Zhicheng GENG, Shengwen TANG, Yang WANG, Hubao A, Zhen HE, Kai WU, Lei WANG, 2024. Stress relaxation properties of calcium silicate hydrate: a molecular dynamics study. *Journal of Zhejiang University-SCIENCE A (Applied Physics & Engineering)*, 25(2):97-115. <u>https://doi.org/10.1631/jzus.A2300476</u>

# Stress relaxation properties of calcium silicate hydrate: a molecular dynamics study

Key words: Calcium-silicate-hydrate; Stress relaxation; Ca/Si ratio;

Temperature; Water content; Atomic simulation

Corresponding author: Shengwen TANG

Email: tangsw@whu.edu.cn

ORCID: https://orcid.org/0000-0002-4883-3103

# Motivition

- 1. The time-dependent viscoelastic response of cement-based materials to applied deformation is far from fully understood at the atomic level.
- 2. Calcium silicate hydrate (C-S-H), the main hydration product of Portland cement, is responsible for the viscoelastic mechanism of cement-based materials.
- 3. The effects of various factors on the stress relaxation properties of C-S-H were systematically investigated by molecular dynamics simulation to shed light on the stress relaxation properties of C-S-H from a microscopic perspective, bridging the gap between the microscopic phenomena and the underlying atomic-level mechanisms.

#### Evolution of stress



Fig. 4 The change in stress of C-S-H subjected to different shear deformations at T = 298 K: (a) evolution of  $\Delta \sigma$  under  $\gamma_{xz}$ ; (b) evolution of  $\Delta \sigma$  under  $\gamma_{yz}$ 

- The stress evolution of the entire C-S-H system takes on a typical L-shape during the stress relaxation process.
- once the initial shear deformation exceeds the elastoplastic range, some micro cracks are probably generated within C-S-H systems, resulting in a rela-tively small Δσ during the stress relaxation process

#### Time correction function



Fig. 7 C(t) of the interlayer Ca-O bonds and H-bonds under different shear states considered here: (a) C(t) of interlayer Ca-O bonds under  $\gamma_{xz}$ ; (b) C(t) of H-bonds under  $\gamma_{xz}$ ; (c) C(t) of interlayer Ca-O bonds under  $\gamma_{yz}$ ; (d) C(t) of H-bonds under  $\gamma_{yz}$ .

#### Mean square displacement



Fig. 11 MSD of constituent parts in the C-S-H system under  $\gamma_{xz} = 4^{\circ}$ : (a) MSD of each atom/molecule/group in C-S-H at T = 298 K; (b) MSD of interlayer calcium; (c) MSD of hydroxyl group; (d) MSD of water molecule.

#### Nonaffine squared displacement



Fig. 15 Intensity plots of nonaffine squared displacement of C-S-H system with different water contents ( $r_{cut} = 3 \text{ Å}$ ).

#### Conclusions

- The stress relaxation response of C-S-H occurs regardless of whether it is under initial shear, tensile, or compressive deformation, and shows a heterogeneous characteristic.
- A large Ca/Si ratio or high temperature leads to low viscosity in the interlayer region and low cohesion between the calciumsilicate layer and the interlayer region, resulting in high atomic dynamics, especially for water molecules, hydroxyl groups, and interlayer calcium atoms.
- Water molecules play a critical role in the morphology and interlayer hydrogen bonding network of C-S-H, which in turn leads to C-S-H exhibiting different stress relaxation properties at different water contents.